

# Chemical Reactions

**Chemical Reaction**: a process that changes one set of chemicals into another.

What is the general format for writing a chemical reaction?

Reactant(s)



Product(s)

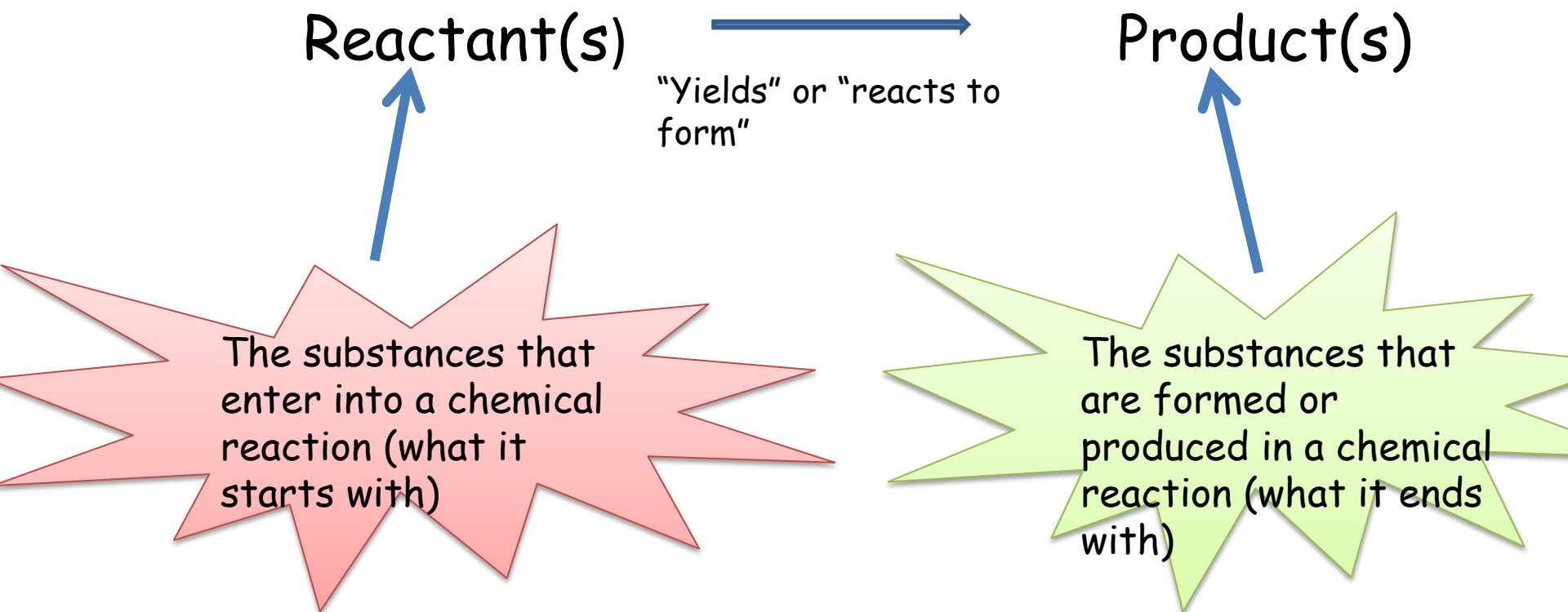
"Yields" or "reacts to form"

The substances that enter into a chemical reaction (what it starts with)

The substances that are formed or produced in a chemical reaction (what it ends with)

# Chemical Reactions

- Chemical reactions always involve **changes in chemical bonds** between atoms.



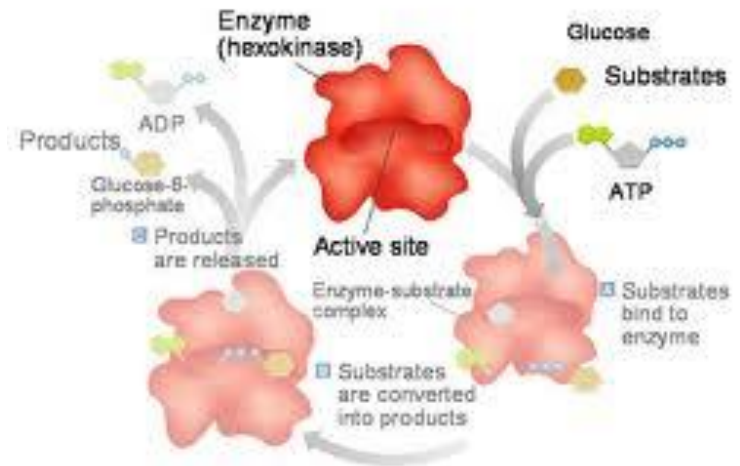
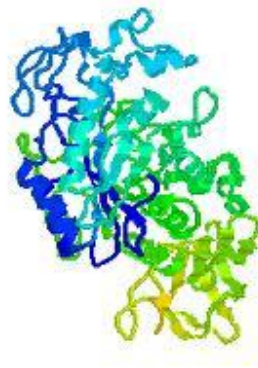
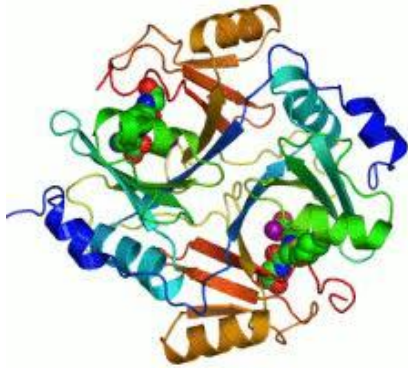
# Enzymes Are Biological Catalysts

**Catalyst** – a substance that speeds up chemical reactions.

- many reactions in the body happen too slowly on their own and need help to speed up.
  - Ex. our cells are constantly producing  $\text{CO}_2$ , it gets dissolved into our bloodstream to carry it away from our tissues.
  - $\text{CO}_2$  doesn't dissolve into blood easily, so enzymes help the blood carry the  $\text{CO}_2$  and then we exhale it.

# Enzymes Are Biological Catalysts

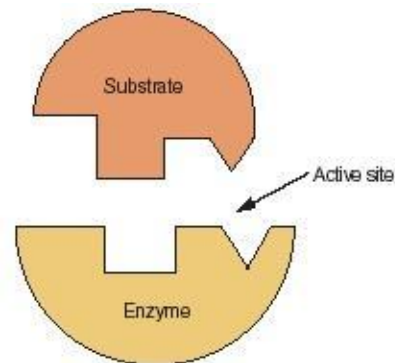
In living things, biological catalysts are known as **enzymes** – they speed up chemical reactions that take place in cells.



**ENZYMES ARE PROTEINS!!!!!!!!!!!!!!!!!!!!!!**

# How Enzymes Work

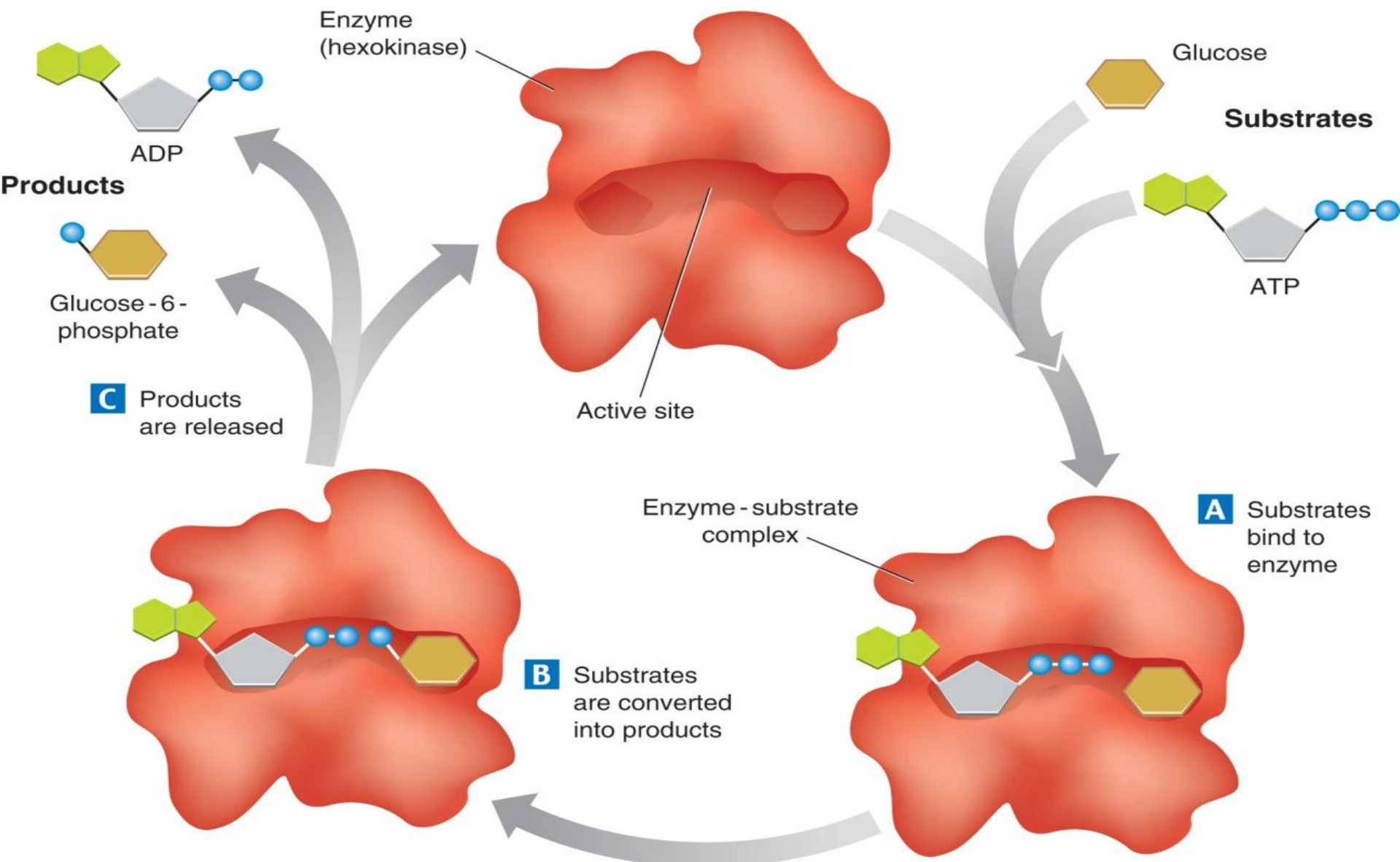
- Enzymes are very specific – they speed up only certain reactions.
- Substrates: the reactants in a reaction
- Active Site: where the substrates bind to the enzyme.
- The **shapes** of the substrates and the active site are **complementary** - they fit together very well (lock and key).



son Prentice



# Enzyme Action

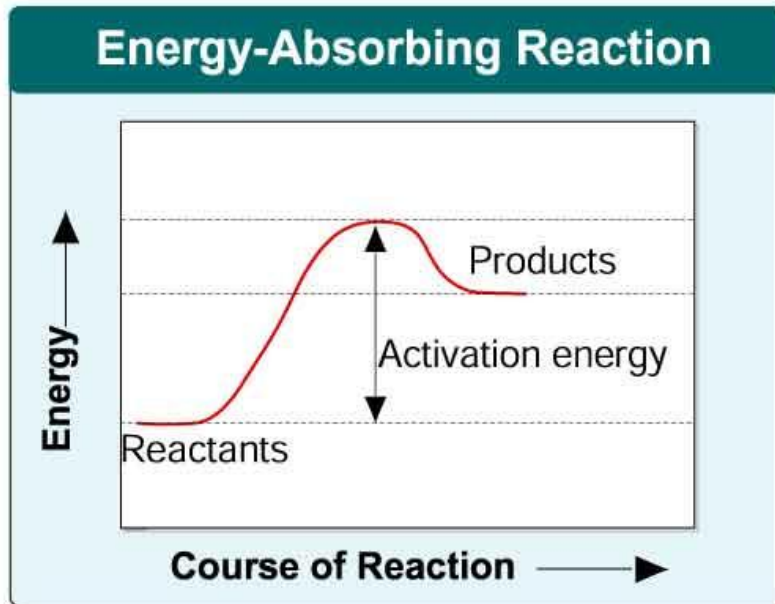


# How Enzymes Work

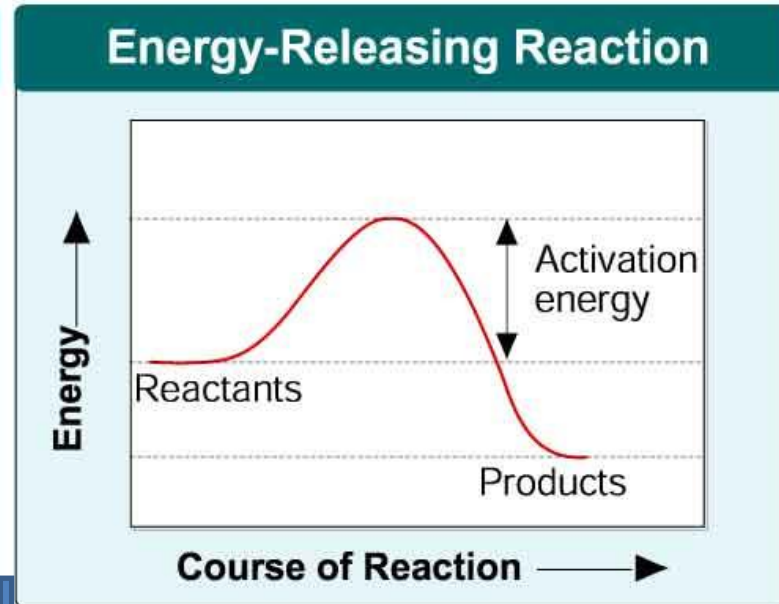
[http://highered.mcgraw-hill.com/sites/0072495855/student\\_view0/chapter2/animation\\_how\\_enzymes\\_work.html](http://highered.mcgraw-hill.com/sites/0072495855/student_view0/chapter2/animation_how_enzymes_work.html)

# Chemical Reactions and Energy

- Some chemical reactions absorb energy (photosynthesis)
- Some chemical reactions release energy (digestion reactions).



The energy of the products is \_\_\_\_\_ than that of the reactants.

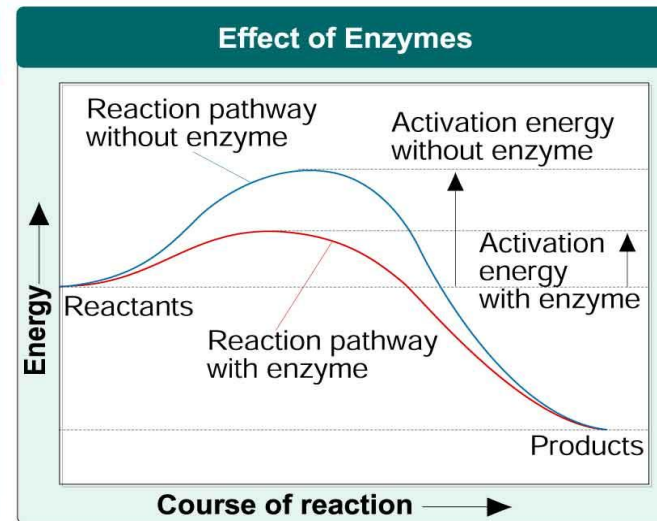
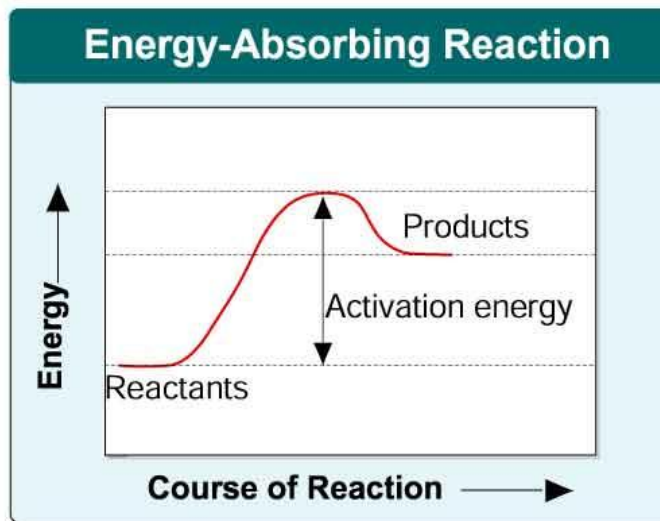


The energy of the products is \_\_\_\_\_ than that of the reactants.



# Chemical Reactions and Energy

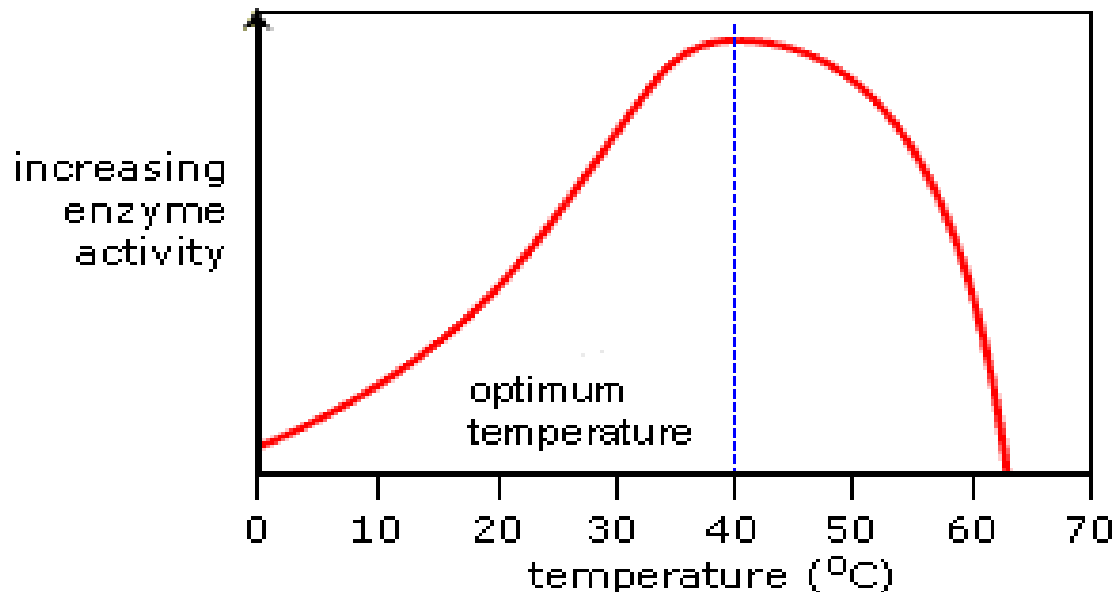
- **Activation energy** – energy needed to get a reaction started.
- Enzymes lower the activation energy – speeds up the reaction.



# Conditions That Effect Enzyme Activity

## Temperature:

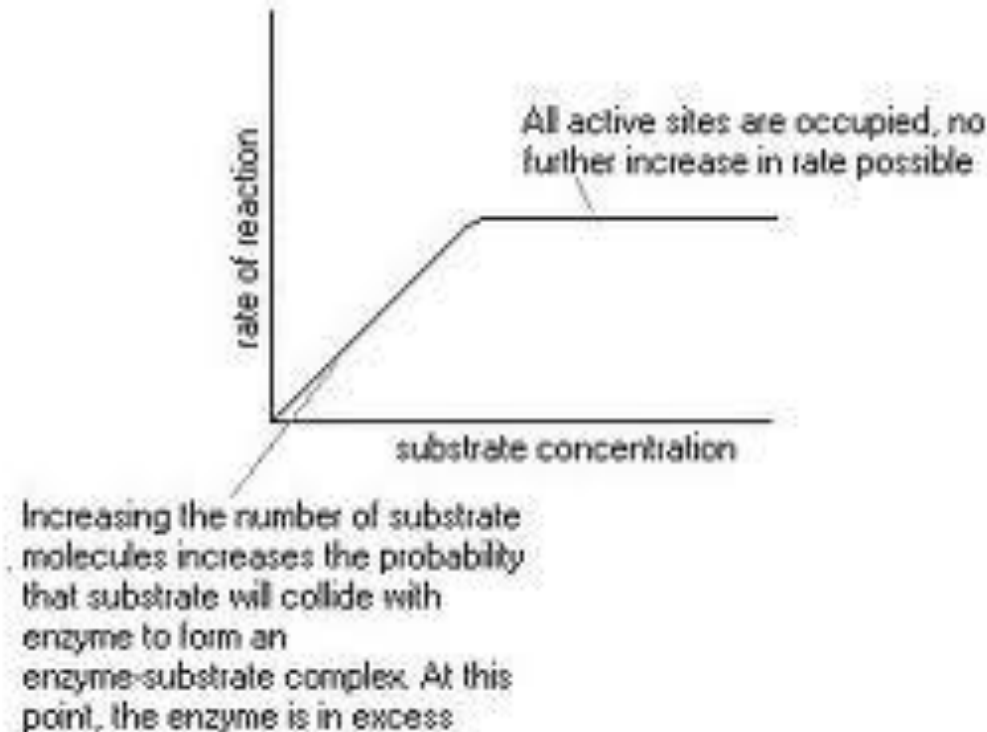
- **Lower** temp = **slower** enzyme activity.
- **Higher** temp = **faster** enzyme activity.
- BUT - if the temperature gets too high, the enzyme **denatures** or **changes shape** so that it works ineffectively, or not at all!



# Conditions That Effect Enzyme Activity

## Concentration of Substrate:

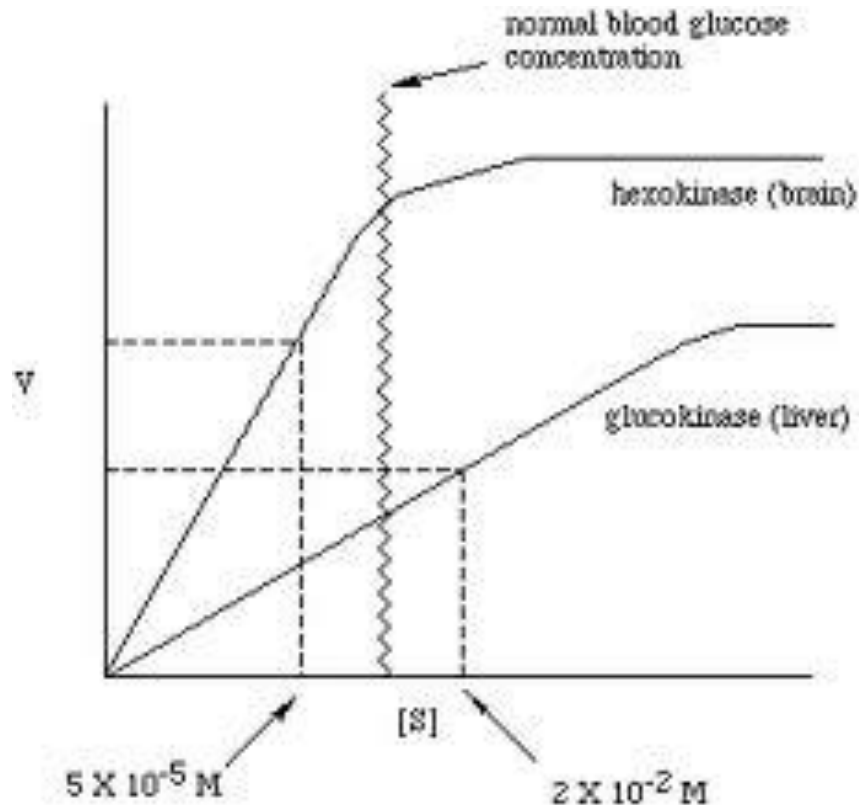
- **Lower** concentration = **slower** enzyme activity.
- **Higher** concentration = **faster** enzyme activity.



# Conditions That Effect Enzyme Activity

## Concentration of Enzyme:

- **Lower** concentration = **slower** enzyme activity.
- **Higher** concentration = **faster** enzyme activity.



# Conditions That Effect Enzyme Activity

- **Inhibitors** – inhibitors stop the substrates from binding to the enzyme's active site.
  - The reaction involving the substrates cannot be sped-up!
  - The rate of reaction decreases!

